PREPARATION AND STRUCTURE OF (<u>R</u>)-(-)- AND (<u>S</u>)-(+)-2, 2'-(2, 2-DIMETHYL-2-SILAPROPANE-1, 3-DIYL)-1, 1'-BINAPHTHALENE

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<u>Summary</u>. The title axially dissymmetric organosilicon compound is obtainable in optically pure form by synthesis of the racemate followed by resolution with a chiral poly(triphenylmethyl methacrylate) column. The structure has been established by spectroscopic and single crystal X-ray diffraction methods.

Axially dissymmetric molecules are of great significance from both stereochemical 1 and preparative $^{2-5}$ point of view. Particularly biaryls which show atropisomerism occupy a central place in this area. Despite recent revolutional development of organosilicon chemistry, 6 chiral organosilicon compounds have been limited to those possessing an asymmetric silicon atom⁷ or a centro-dissymmetric alkyl substituent. We have prepared for the first time an axially dissymmetric tetraalkylsilane in a chiral form.

Synthesis of the racemic organosilicon compound, 2, possessing a \underline{C}_2 axis is simply based on the condensation of a dilithio derivative of (\pm) -2, 2'-dimethyl-1, 1'-binaphthyl (1) and a dichlorosilane. Thus treatment of (\pm) -1 with 3 equiv of <u>n</u>-butyllithium/<u>N,N,N',N'</u>-tetramethylethylenediamine in hexane (-20 °C, 24 h) followed by exposure to 3 equiv of dichlorodimethylsilane (0 °C, 1 h) produced the desired (\pm) -2, mp 163-165 °C, in 50% yield. The



spectral and analytical data were consistent with the assigned structure.⁸

This organosilane possesses only aliphatic and aromatic hydrocarbon groupings and lacks obvious functionalities for optical resolution. Fortunately, however, the recently developed chiral poly(triphenylmethyl methacrylate) (PTrMA) column⁹ provides a powerful tool for optical resolution of racemates of this type. Thus when (\pm) -2 was subjected to high-pressure liquid chromatography (HPLC) on a column of (+)-PTrMA coated on macroporous silica gel (Chiralpak OT)¹⁰ by using methanol as eluant, the complete resolution was attained (Figure 1).¹¹ Preparative separation followed by plug filtration through silica gel and recrystallization from methanol afforded analytical sample of (+)-2 and (-)-2 as colorless prisms. The circular dichroism (CD) curve of the levorotatory enantiomer, ¹² mp 169-171 °C, $[\alpha]_D^{22}$ -324° (<u>c</u> 0.089, dioxane), resembled closely those of the known related (<u>R</u>)-2, 2'- bridged 1, 1'-binaphthyl derivatives¹³ (Figure 2), and thus assigned to have the <u>R</u> configuration.



Figure 1. HPLC chromatogram for the resolution of racemic 2 on a Chiralpak OT column.



Figure 2. CD and UV spectra of $(\underline{\mathbf{R}})$ -(-)-2 (dioxane).

The molecular structure of the levorotatory 2 has been determined by the X-ray crystal structure analysis (Figure 3).¹⁴ The absolute configuration was firmly established to be <u>R</u>. Some bond lengths and angles of interest are listed in Table I. The mean S_1-CH_3 bond length [1.861(4) Å] is normal, ¹⁵⁻¹⁷ but the lengths of the $S_1-C(11)$ and $S_1-C(11')$ bonds [1.884(3) and 1.893(3) Å, respectively] are rather long and the $C(11)-S_1-C(11')$ angle is

narrowed to 103.7(1)°. The angle between the least-squares planes through the two naphthyl groups is 65.5°.¹⁸



Figure 3. Stereoscopic view of (R)-(-)-2.

$S_1 - C(M)$	1,861(3)	$C(M) - S_1 - C(M')$	110.4(2)
$S_1 - C(M')$	1.861(4)	$C(M) - S_1 - C(11)$	110.8(1)
$S_1 - C(11)$	1.884(3)	C(M)-S1-C(11')	111.0(2)
S1-C(11')	1.893(3)	C(11)-S1-C(11')	103.7(1)
C(2)—C(2')	1.492(3)	$C(M') - S_1 - C(11)$	109.4(1)
		$C(M') - S_1 - C(11')$	111.4(1)

Table I. Selected Bond Lengths (A) and Angles (Deg) for (\underline{R}) -(-)-2

Obviously the achiral synthesis/resolution sequence described herein allows an easy entry to various organic and organometalloid compounds having related chiral structures. Further application of this technique is in progress.

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- 8. ¹H NMR (benzene with tetramethylsilane) δ -0.10 (s, 6H, CH₃), 1.59 and 1.82 (ABq, $J_{AB} = 13$ Hz, 4H, CH₂), 7.8-8.8 (m, 12H, aromatic). C NMR (CDCl₃, solvent signal taken as $\delta = 77.1$) δ 3.5 (CH₃), 23.9 (CH₂), 124.2, 125.8, 126.5, ³127.8, 128.0, 131.8, 132.3, 132.3, 132.8, 137.1 (aromatic). IR (CCl₄) 3050-2900, 1640, 1580, 1510 cm⁻¹. UV^{dloxane} ($\varepsilon \times 10^{-4}$) 227 (9.26), 237 (8.48), ³302 (0.97), 315 (0.67, shoulder), 329 (0.49) max nm. MS m/z 338 (M⁺). Anal. Calcd for C₂₄H₂₂Si: C, 85.15, H, 6.55. Found: C, 84.93, H, 6.74.
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- Optically active, isotactic PTrMA coated on LiChrospher Si 1000. Commercially available from Japan Spectroscopic Co. Ltd., Ishikawa-cho, Hachioji, Tokyo, Japan.
- Waters 6000A instrument equipped with a JASCO UVIDEC-100 ultraviolet detector (292 nm). Column, 25 cm x 0.46 cm; eluant, methanol, flow rate, 1 mL/min, temp, 16,°C.
- 16 °C. 12. $\lambda^{\text{dioxane}}_{\lambda}$ ($\Delta \epsilon$) 340 (0.0), 329 (+10.4), 325 (+6.0), 314 (+14.1, shoulder), 307 (+20.2, shoulder), 301 (+21.9), 289 (+19.8), 287 (+20.1), 263 (-14.4, shoulder), 254 (-16.5, shoulder), 240 (-558), 224 (+395), 210 (+56.5) nm, (c 3.54 x 10⁻⁵).
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- 14. The crystals are colorless prisms. Intensities of the reflections were measured on a Rigaku AFC-5 diffractometer with Cu K α radiation up to $2\theta = 135^{\circ}$. The crystallographic data are SiC₂₄H₂₂, ${}_{\circ}$ M_r = 338.52, orthorhombic, P2₁2₁2₁, a = 12.441(1), b = 20.331(1), c = 7.581(1) A, V = 1917.5(3) A, D = 1.173 g cm⁻³, Z = 4, and μ (Cu K α) = 21.21 cm⁻¹. The structure was solved by a direct method (MULTAN) and refined by block-diagonal least-squares calculations including hydrogen atoms. The final R and R values were 0.040 and 0.049, respectively, for a total of 3089 (1783 hkl and 1306 hkl)^W reflections with $|Fo| > 3\sigma(|Fo|)$. The enantiomeric structure refined under identical conditions gave R and R values of 0.046 and 0.057, respectively. The R configuration was also confirmed by comparison of the observed intensity relations between hkl and hkl reflections with the calculated relations. X-ray data (Tables of atomic coordinates, temperature factors, structure factors, bond lengths, and angles) have been deposited with the Cambridge Crystallographic Data Centre.
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- 18. Correspondence about X-ray crystallography should be made to T.I. among the authors.

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